

# Fe-Ho-Ni (Iron-Holmium-Nickel)

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[1993Fan] determined an isothermal section at  $\sim 25^\circ\text{C}$  for this system for Ho contents up to 33.3 at.%.

## Binary Systems

The Fe-Ho phase diagram reviewed by [1982Kub] depicts four line compounds:  $\text{Fe}_{17}\text{Ho}_2$ ,  $\text{Fe}_{23}\text{Ho}_6$ ,  $\text{Fe}_3\text{Ho}$ , and  $\text{Fe}_2\text{Ho}$ . See [Massalski2] for the Fe-Ni phase diagram. Iron and nickel form a complete solid solution (fcc,  $\gamma$ ) at high temperatures, which decomposes eutectoidally at  $347^\circ\text{C}$  to ( $\alpha\text{Fe}$ ) and an ordered phase  $\text{FeNi}_3$ . The Ho-Ni phase diagram was determined by [1991Zho]. There are eight compounds in this system:  $\text{Ho}_3\text{Ni}$ ,  $\text{Ho}_3\text{Ni}_2$ ,  $\text{HoNi}$ ,  $\text{HoNi}_2$ ,  $\text{HoNi}_3$ ,  $\text{Ho}_2\text{Ni}_7$ ,  $\text{HoNi}_5$ , and  $\text{Ho}_2\text{Ni}_{17}$ . See [Pearson3] for structural data on the binary compounds.

## Ternary Isothermal Section

With starting metals of purity of 99.95% Fe, 99.95% Ho, and 99.99% Ni, [1993Fan] prepared 123 alloy samples by induction melting in an Ar atm. The alloys were annealed at  $500^\circ\text{C}$  for 5 days and cooled slowly at the rate of  $10^\circ\text{C h}^{-1}$  to room temperature ( $\sim 25^\circ\text{C}$ ). It is presumed that the phase

equilibria correspond to  $\sim 25^\circ\text{C}$ . The phase identification was carried out by x-ray powder diffraction and electron probe microanalysis techniques. Their isothermal section at  $\sim 25^\circ\text{C}$  is redrawn in Fig. 1 to agree with the accepted binary data. [1993Fan] did not report the existence of the  $\text{FeNi}_3$  phase. The homogeneity ranges of ( $\alpha\text{Fe}$ ),  $\text{FeNi}_3$ , and  $\gamma$  shown in Fig. 1 are approximate. The  $\text{PuNi}_3$ -type isostructural compounds  $\text{Fe}_3\text{Ho}$  and  $\text{HoNi}_3$  form a complete series of solid solutions  $\text{Ho}(\text{Fe,Ni})_3$ , denoted 1:3 in Fig. 1.  $\text{Fe}_{17}\text{Ho}_2$  and  $\text{Fe}_{23}\text{Ho}_6$  dissolve about 10 and 7 at.% Ni, respectively, at constant Ho content.  $\text{HoNi}_5$  and  $\text{HoNi}_2$  dissolve about 24 and 23 at.% Fe at constant Ho content. The lattice parameter of the cubic Laves phase  $\text{HoNi}_2$  varies linearly from 0.7160 nm at 0% Fe to 0.7244 nm at 23 at.% Fe. The Fe-Ni phases dissolve about 2 at.% Ho. The 1:3 phase comes into equilibrium with a phase  $\chi$  of high Ho content. The  $\chi$  phase showed some weak diffraction lines similar to those of  $\text{Ho}_3\text{Ni}_2$  [1993Fan]. More experiments are needed to determine its composition and structure.

## References

- 1982Kub:** O. Kubaschewski: *Iron-Binary Phase Diagrams*, Springer-Verlag, Berlin, 1982, pp. 111-12.  
**1991Zho:** H. Zhou, Y. Ou, and X. Zhong: *J. Alloys Compounds*, 1991, vol. 177, pp. 101-06.  
**1993Fan:** X. Fang, H. Zhou, X. Ou, and Y. Zhuang: *J. Alloys Compounds*, 1993, vol. 196, pp. L15-L17.

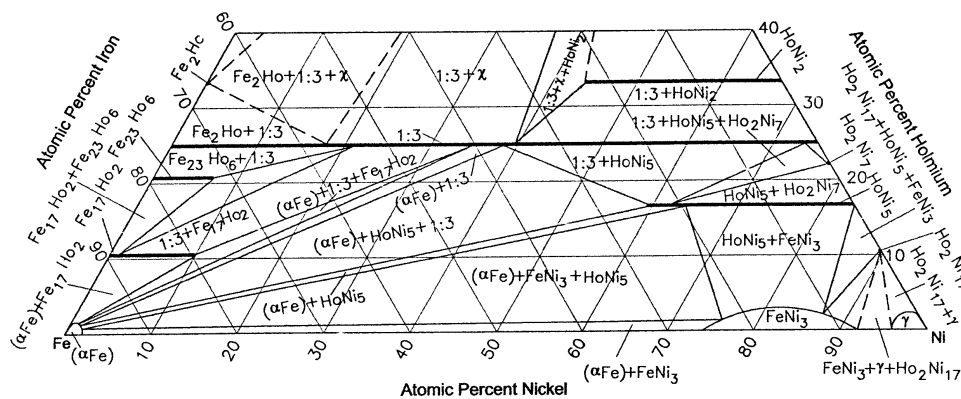


Fig. 1 Fe-Ho-Ni partial isothermal section at  $\sim 25^\circ\text{C}$  [1993Fan]